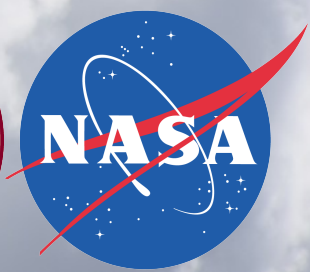


# Heterogeneous cloud and aerosol chemistry in the tropospheric $\text{NO}_y$ cycle

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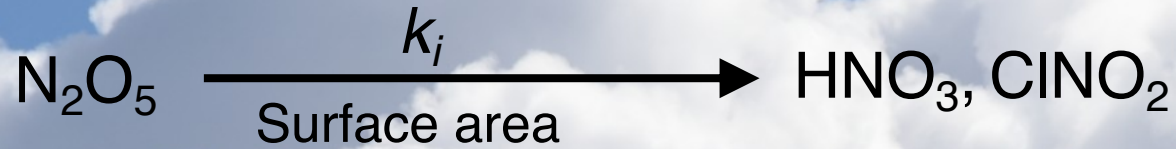
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# $\text{N}_2\text{O}_5$ hydrolysis on aerosol surfaces: a large $\text{NO}_x$ sink

Field: Platt et al., 1980, 1984; Heikes and Thompson, 1983; Brown et al., 2007

Models: Dentener & Crutzen, 1993; Tie et al., 2001; Evans and Jacob, 2005  
*and many more*



Particle size

Diffusion

Reactive uptake  $\gamma$

(Schwartz, 1986; Sander, 1999; Jacob, 2000)

Aerosol:  $\gamma \approx 10^{-4} - 0.02$  (complicated)

Water & Ice:  $\gamma \approx 0.02 - 0.03$  (well known)

(Crowley et al., 2010; Ammann et al., 2013;  
Burkholder et al., 2015)

- **20-35% of global  $\text{NO}_x$  sink** (Alexander et al., 2009; Bauer et al., 2004)
- **Reduces global  $\text{O}_3$  and OH by 1-4%** (Macintyre and Evans, 2010)

## *What about clouds?*

**Clouds have  $10^6$  times greater surface area than aerosols** (Holmes et al., 2019)

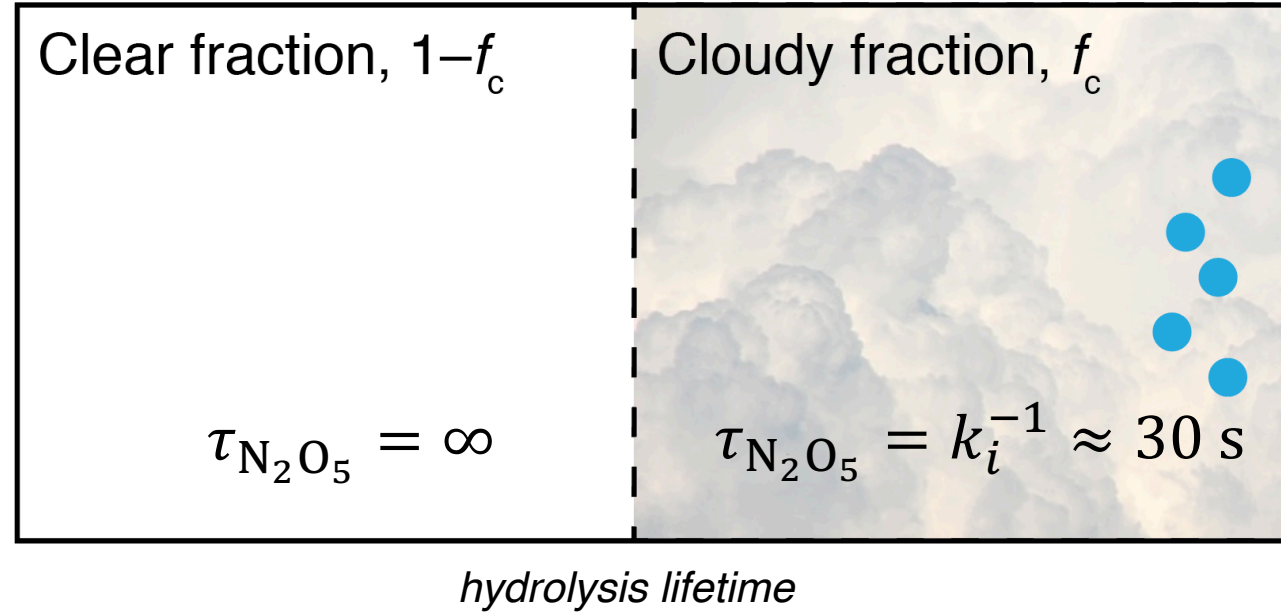
...but cloud contact is sporadic, roughly daily (Lelieveld et al., 1989)

**Past studies reported cloud uptake is very small, because they assumed aerosol uptake was very fast ( $\gamma = 0.1$ )** (Lelieveld and Crutzen, 1990; Dentener & Crutzen, 1993; Jacob, 2000)

# Numerical methods for cloud uptake in regional and global models

\*that don't resolve clouds

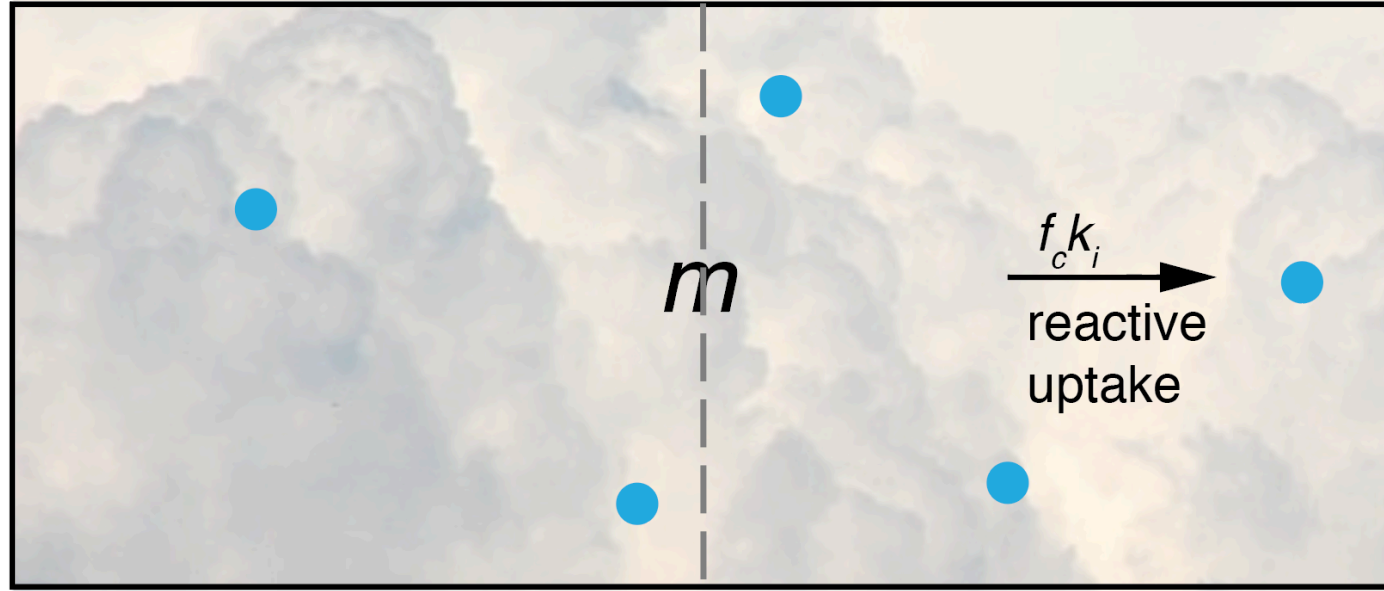
Grid cells are *partly* cloudy, usually  $f_c < 0.2$



# Numerical methods for cloud uptake in regional and global models

## Existing methods

### Thin-cloud approximation



**Problem:** uptake much too fast



# Numerical methods for cloud uptake in regional and global models

## Existing methods

Thin-cloud approximation

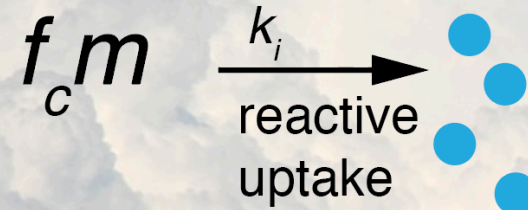


## Repartitioning method

Clear fraction,  $1-f_c$

$$(1-f_c)m$$

Cloudy fraction,  $f_c$

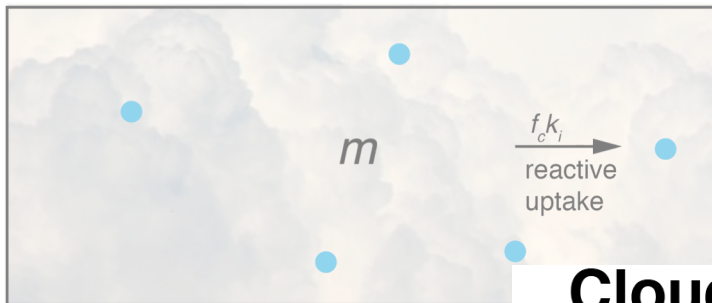


**Problem:** sensitive to model time stepping (unphysical)  
converges to thin-cloud approximation

# Numerical methods for cloud uptake in regional and global models

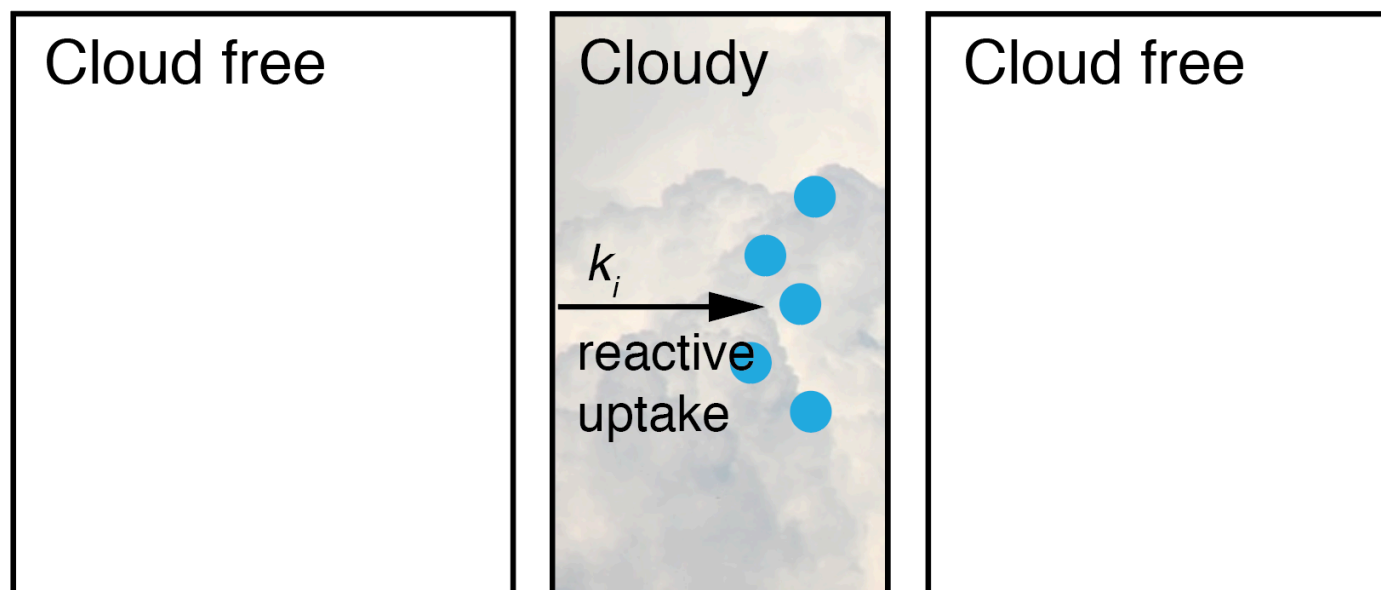
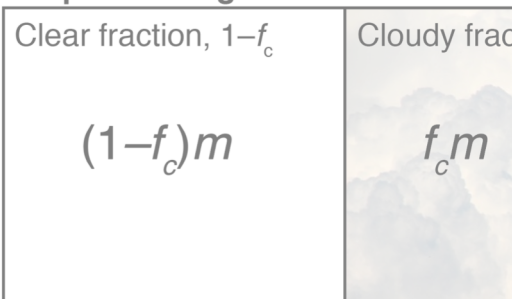
## Existing methods

### Thin-cloud approximation



## Cloud pulsing

### Repartitioning method



**Problem:** chemical operator splitting  
complicated  
not used in any current models

Time

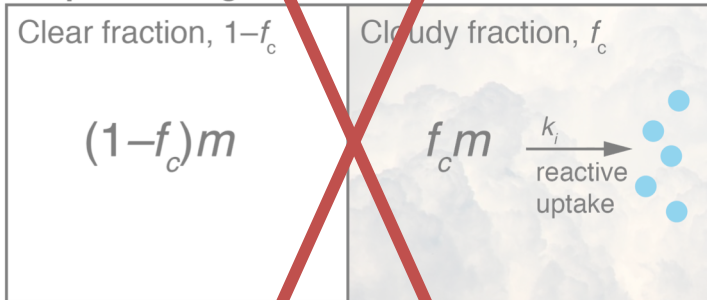
# Numerical methods for cloud uptake in regional and global models

## Existing methods

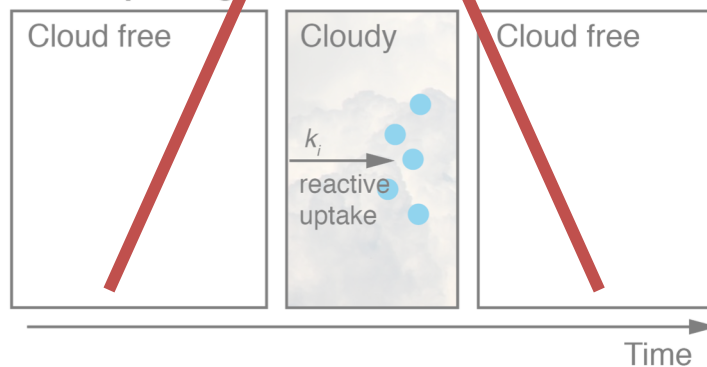
### Thin-cloud approximation



### Repartitioning method



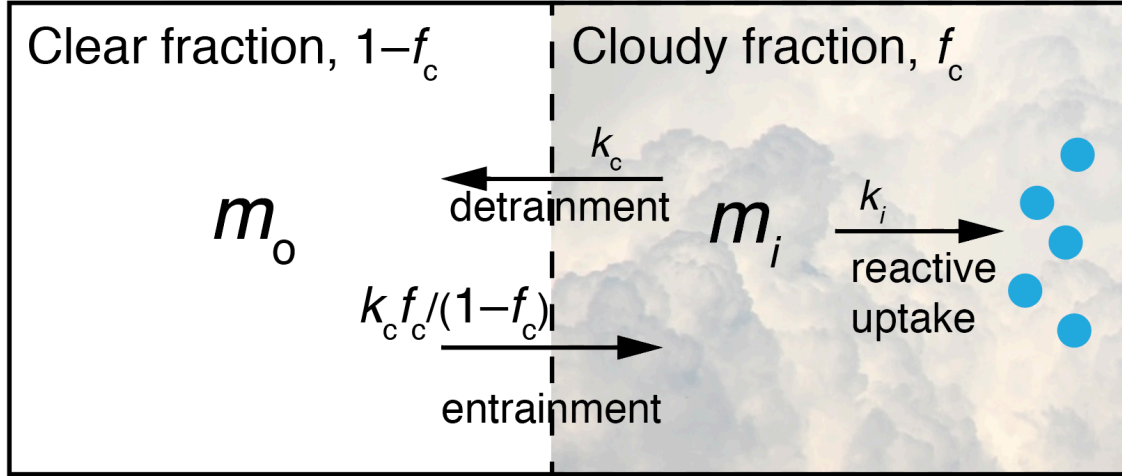
### Cloud pulsing



# Numerical methods for cloud uptake in regional and global models

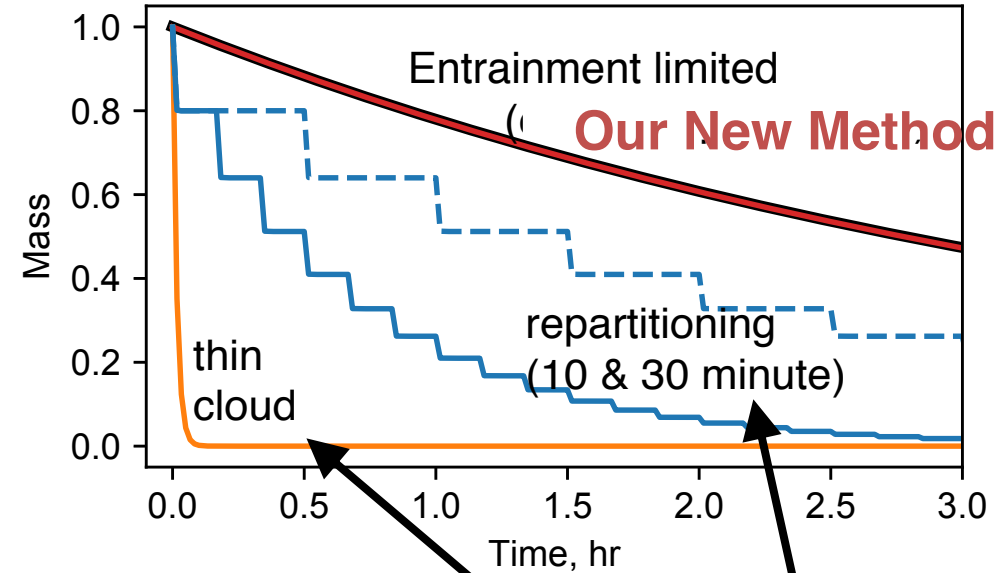
## New method

### Entrainment-limited uptake



A single, 1<sup>st</sup> order loss rate integrating in-cloud uptake, cloud fraction, and entrainment

### Example errors



### Partly cloudy, entrainment limited (new):

$$\tau_{\text{N}_2\text{O}_5} \approx \left( \underbrace{\frac{1 - f_c}{f_c k_c}}_{\text{Entrainment}} + \underbrace{\frac{r}{f_c A D_g}}_{\text{In-cloud uptake (Schwartz, 1986)}} + \underbrace{\frac{4}{f_c A v \gamma}}_{\text{In-cloud uptake (Schwartz, 1986)}} \right) \sim 10 \text{ hr}$$

Entrainment

In-cloud uptake (Schwartz, 1986)

Methods in current models, including GEOS-Chem

Holmes et al., 2019 GRL

**Advantages:** Physically realistic

Analogous to Schwartz (1986) mass transfer

Simple, fast

Applies to *any* irreversible reaction in clouds:  $\text{N}_2\text{O}_5$ ,  $\text{S(IV)}$ ,  $\text{HO}_2$ ,  $\text{Hg(II)}$ , halogens, aerosol



# Experimental design with GEOS-Chem

## Model configuration, v11-01

Tropospheric chemistry (Parrella et al., 2012)

includes aerosol heterogeneous chemistry (Evans and Jacob, 2005)

MERRA-2 meteorology for 2015 at  $4^\circ \times 5^\circ \times 47\text{L}$

## Model improvements

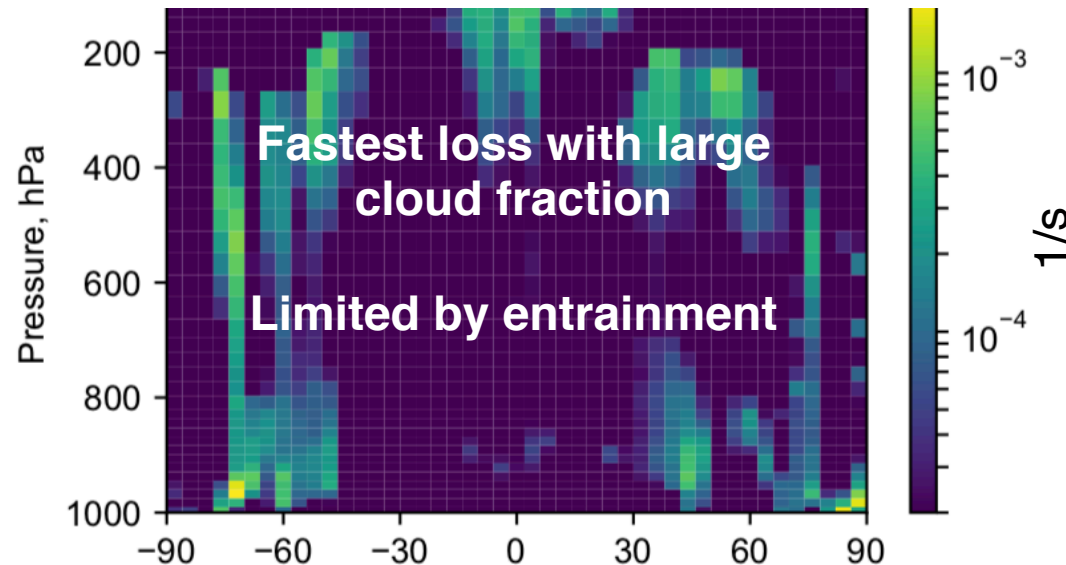
Reassessed  $\gamma$  for all  $\text{NO}_y$  species (JPL, IUPAC)

Uptake on  $\text{SO}_4^{2-}$ – $\text{NO}_3^-$ – $\text{NH}_4^+$  aerosol follows

Bertram and Thornton (2009), Shah et al. (2018)

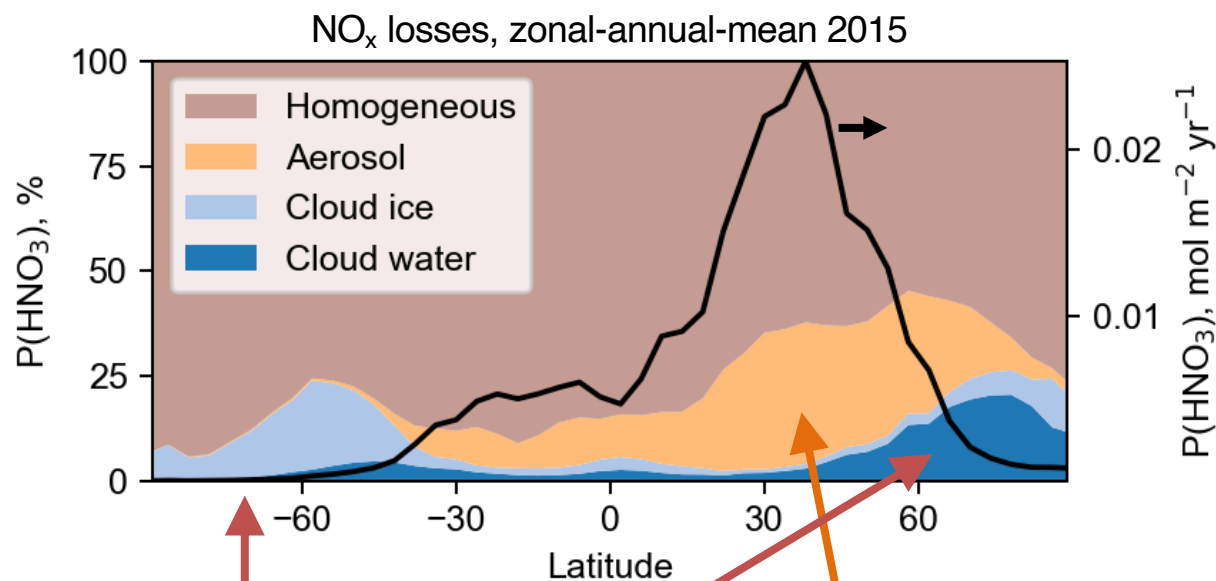
Entrainment-limited cloud uptake

### Entrainment-limited, grid-scale $\text{N}_2\text{O}_5$ loss



MERRA-2 zonal mean, 2015-04-01

# NO<sub>x</sub> losses in improved model



Cloud uptake provides 25% of NO<sub>x</sub> loss in high latitudes (new result)

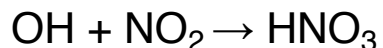
Aerosol uptake provides 30% of NO<sub>x</sub> loss in NH mid latitudes (similar to past literature)

## Global NO<sub>x</sub> losses

Homogeneous

4.1 Tmol(N) yr<sup>-1</sup>

63%



Aerosol uptake

25%

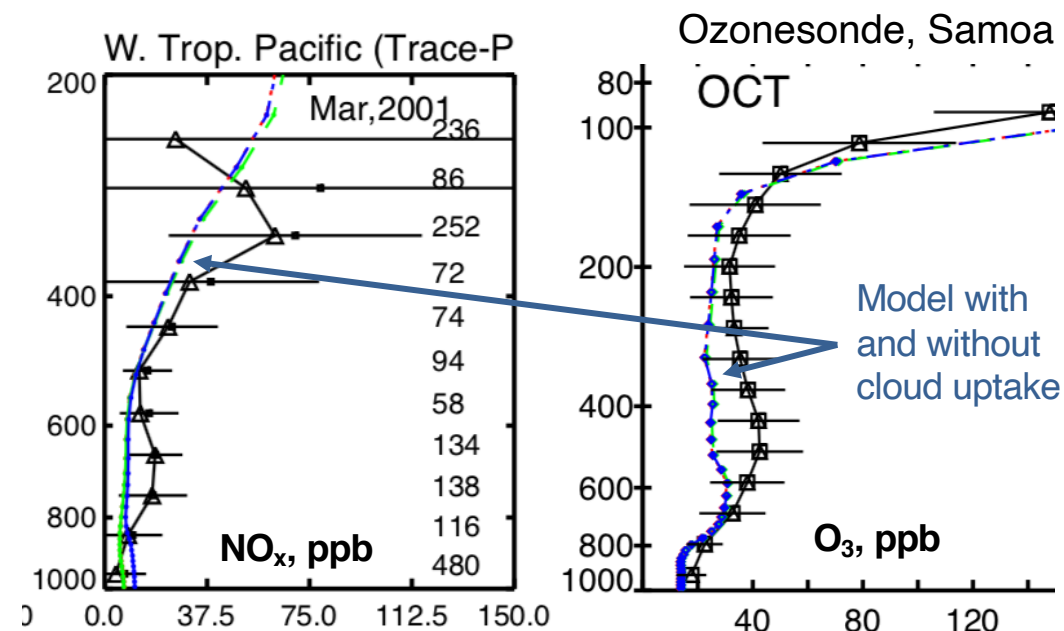
Dry deposition

7%

Cloud uptake

5%

## Model changes are small relative to NO<sub>x</sub> & O<sub>3</sub> environmental variability



Reduces aerosol nitrate bias & improves  $\Delta^{17}\text{O}$ -nitrate  
(Alexander et al., 2019 ACP)

Chemistry is clear: N<sub>2</sub>O<sub>5</sub> hydrolysis should be included in models regardless of performance benefit

# Effects of clouds and aerosol on global tropospheric chemistry

Simulation	Control	Cloud Off	Aerosol Off	Both Off
Aerosol reactions	Yes	Yes	No	No
Cloud reactions	Yes	No	Yes	No
NO <sub>x</sub>	43 Gmol N	+0.7 %	+3.8 %	
O <sub>3</sub>	6.9 Tmol	+2.4 %	+2.5 %	
$\tau_{\text{CH}_4+\text{OH}}$	8.7 yr	-1.9 %	-3.4 %	

## Neglecting cloud uptake of N<sub>2</sub>O<sub>5</sub> and NO<sub>3</sub>

- Increases tropospheric NO<sub>x</sub> by ~1%
- Increases tropospheric O<sub>3</sub> by 2.4%
- Decreases CH<sub>4</sub> lifetime by 2%

Cloud and aerosol heterogeneous reactions have similar magnitude effects on NO<sub>x</sub>, O<sub>3</sub>, and OH

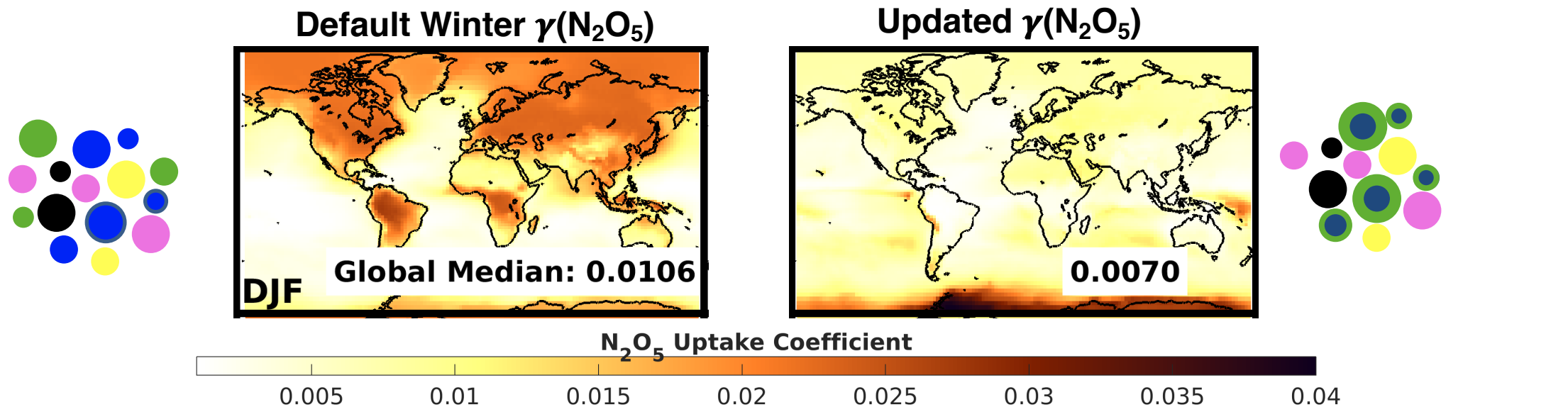
Neglecting both causes larger changes than the sum of their individual effects because clouds and aerosol compete for N<sub>2</sub>O<sub>5</sub>

Accounting for clouds makes tropospheric NO<sub>x</sub>, O<sub>3</sub>, and OH much less sensitive to aerosol  $\gamma$  and surface area.

# Simulating $\text{N}_2\text{O}_5$ uptake on aerosol based on field studies

## McDuffie et al. (2018) empirical parameterization of $\gamma(\text{N}_2\text{O}_5)$

- Internal mixing of organic and sulfate-nitrate-ammonium aerosol
- Accounts for organic coating (Riemer et al., 2009)  
Surrounding an inorganic core (Bertram and Thornton, 2009)
- Better describes WINTER field observations than existing formulas
- Released in GEOS-Chem 12.6



Default GEOS-Chem pre-12.6  
(Evans & Jacob GRL 2005)

Internal mixture of Org+SNA  
(McDuffie et al., JGR 2018)



# Combined changes in tropospheric composition

GEOS-Chem	12.1	+Holmes 2019	+McDuffie 2018	+Holmes 2019 +McDuffie 2018
Aerosol reactions	Yes	Updated $\gamma$	Updated $\gamma$ Organic coating	Updated $\gamma$ Organic coating
Cloud reactions	No	Yes	No	Yes
NO <sub>x</sub>	41 Gmol N	- 0.9 %	+0.5 %	- 0.6 %
Aerosol nitrate	6.9 Gmol N	-1.2 %	- 0.5 %	-1.7 %
O <sub>3</sub>	8.0 Tmol	- 5.2 %	+0.7 %	- 4.9 %
$\tau_{\text{CH}_4+\text{OH}}$	9.4 yr	- 6.1 % (10.0 yr)		- 5.6 % (9.9 yr)

Cloud and internal mixing  
updates partially offset each other

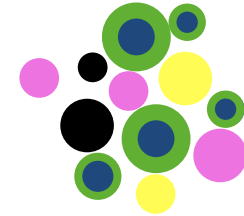
**CH<sub>4</sub> lifetime ( $\tau_{\text{CH}_4+\text{OH}}$ ) now falls within  
observational constraints:  $11.2 \pm 1.3$  yr  
(Prather et al., 2013)**

# Implications for global atmospheric chemistry

**Cloud uptake of  $\text{NO}_x$  plays a significant role** in tropospheric chemistry:

- 20-25% of  $\text{NO}_x$  loss at high latitudes
- 5% of global  $\text{NO}_x$  loss.

This has been overlooked because early studies overestimated aerosol  $\gamma$



After accounting for cloud effects, **aerosols have less (half) influence on tropospheric chemistry**

**Organic coatings on inorganic aerosol** decrease global mean aerosol nitrate and increase  $\text{NO}_x$  by 0.5% each

**Entrainment-limited uptake** is realistic, simple and suitable for other cloud reactions in regional and global chemistry models:  $\text{SO}_2$  oxidation, Halogen recycling,  $\text{HO}_2$  uptake, aerosol processing, aqueous Hg reduction, wet scavenging

