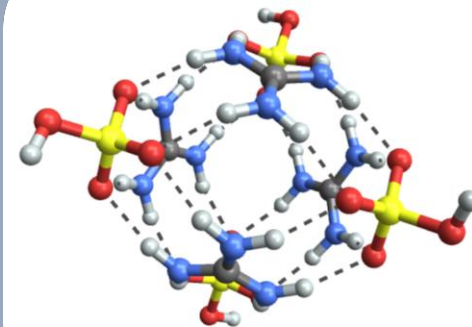


Quantum Chemical Modelling of Atmospheric Molecular Clusters

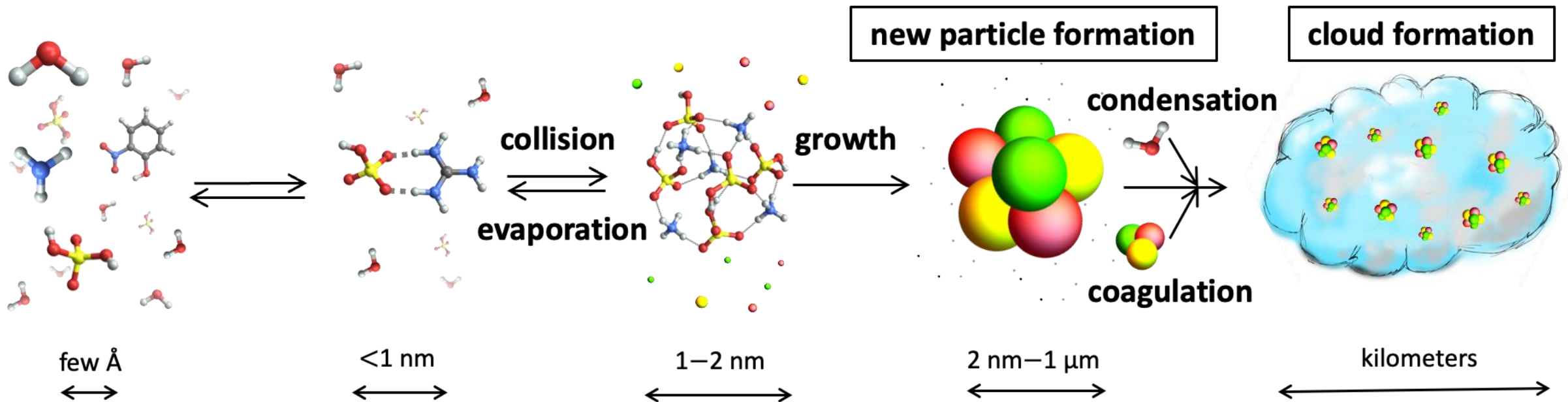


Enhanced by Machine Learning

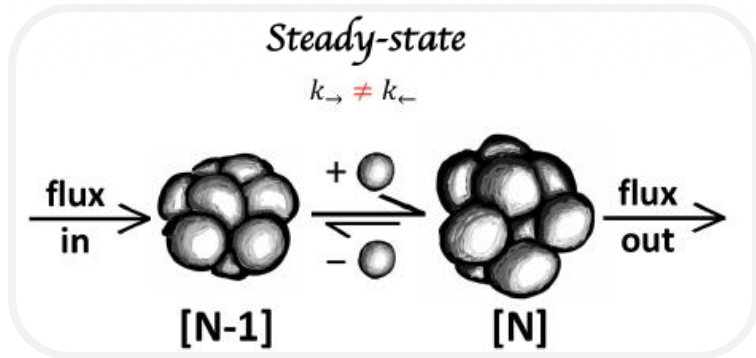
Jakub Kubečka
Aarhus University
Denmark **DK**

New Particle Formation (NPF)

- Aerosols - tiny particles dispersed in the air
 - source of uncertainty in climate modelling [IPCC report 2021]
 - **Primary aerosols** (emitted to the atmosphere, e.g. soot, pollen)
 - **Secondary aerosols** (gas-to-particle formation = NPF)



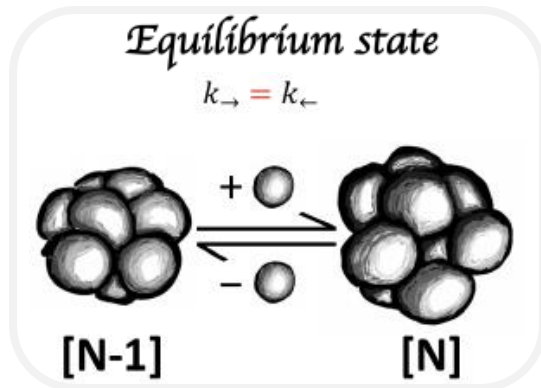
How do we model NPF?



$k_{\rightarrow} \equiv \beta$: collision rate
 (from kinetic gas theory)

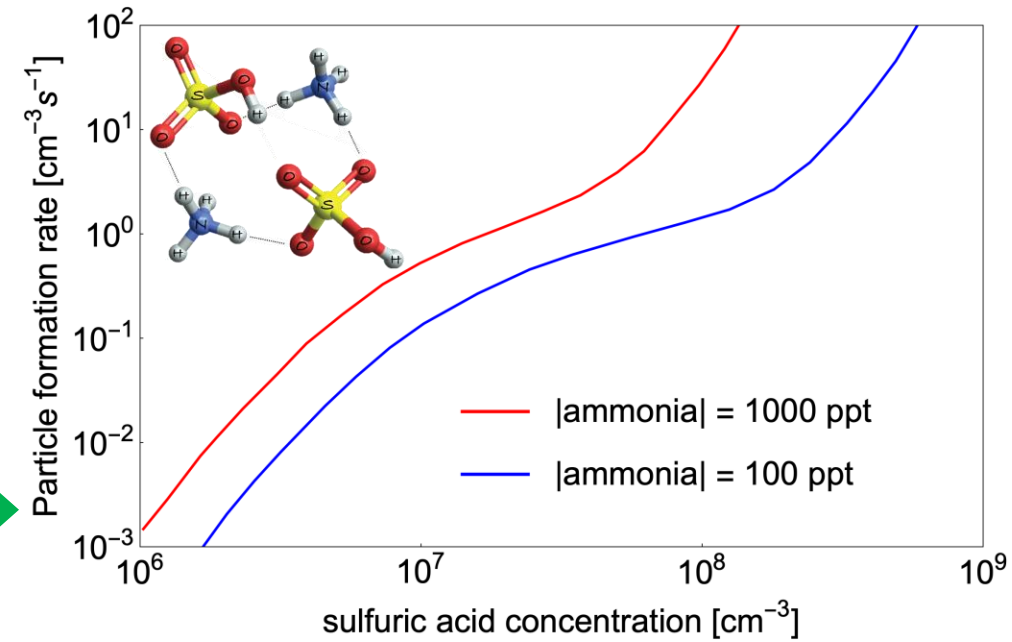
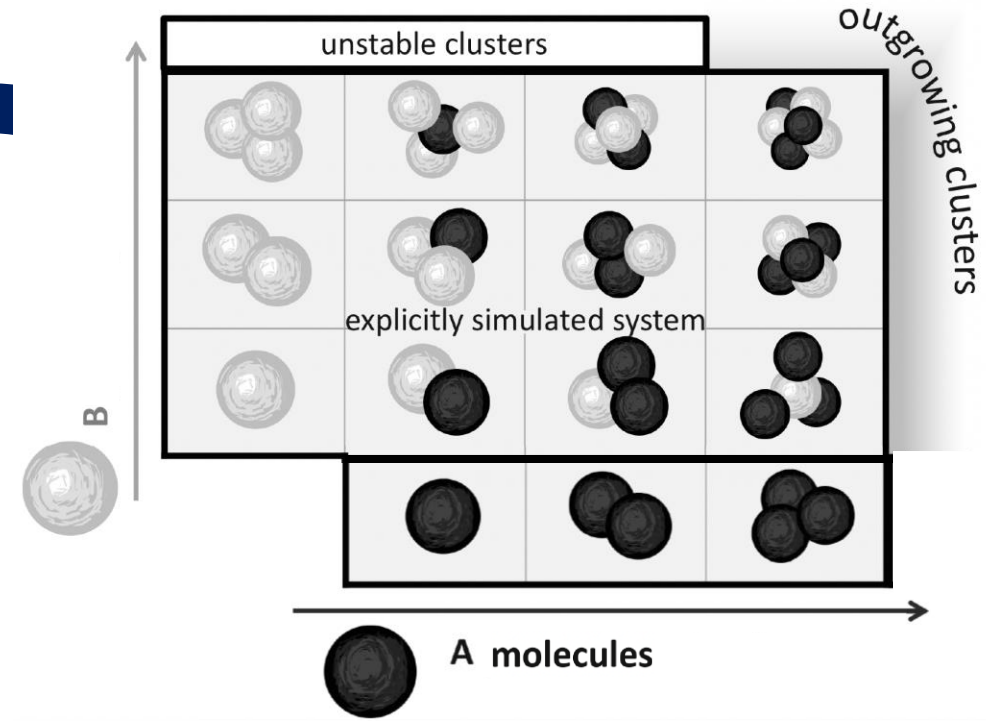
$k_{\leftarrow} \equiv \gamma$: evaporation rate
 (from stat. thermodynamics)

$\gamma = \beta \cdot p \cdot e^{-\Delta G/kT}$



Atmospheric Cluster Dynamics Code

AC/DC

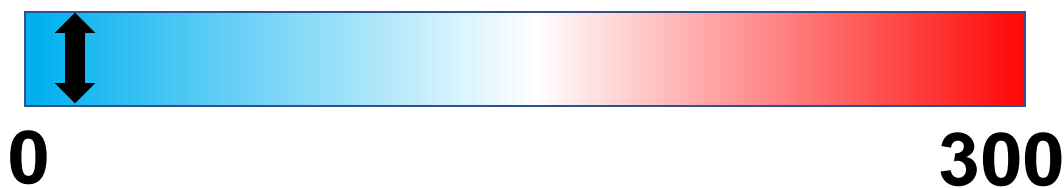
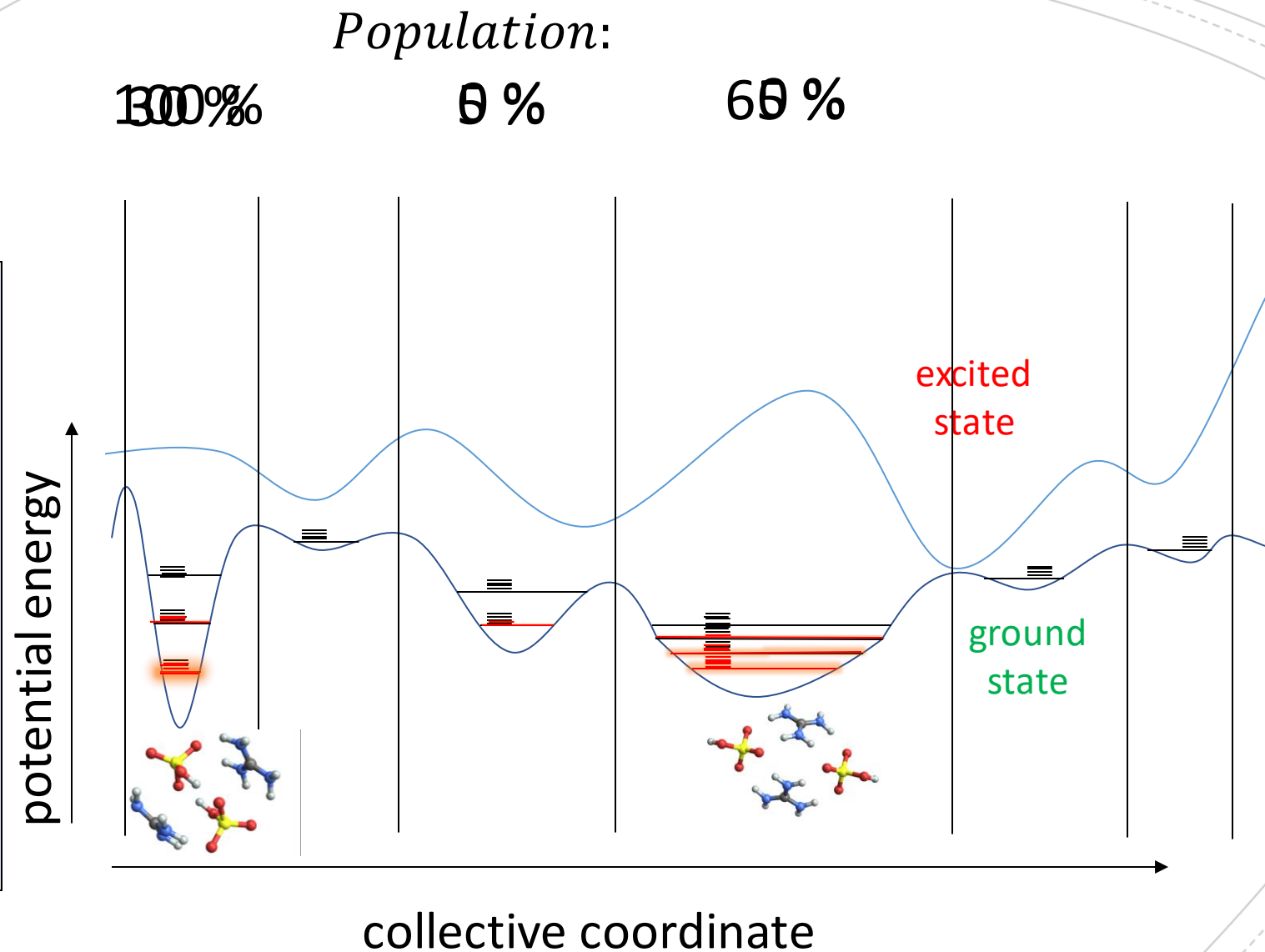


Configurational sampling

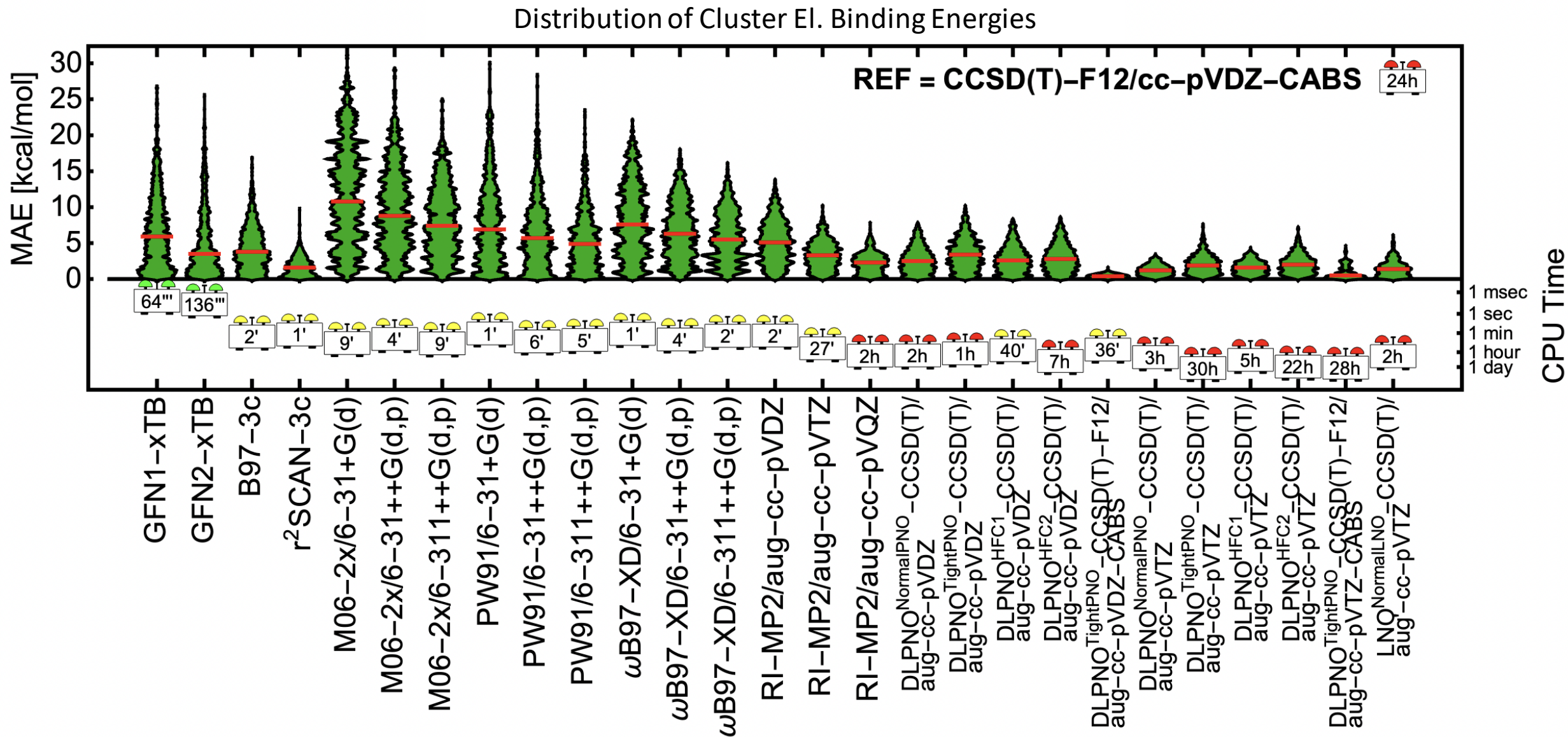
[Kubečka *et al.*, JPC, 2018]

Computational Tools for Molecular Clusters

[Kubečka *et al.*, ACS Ω , 2023]



Benchmarking methods

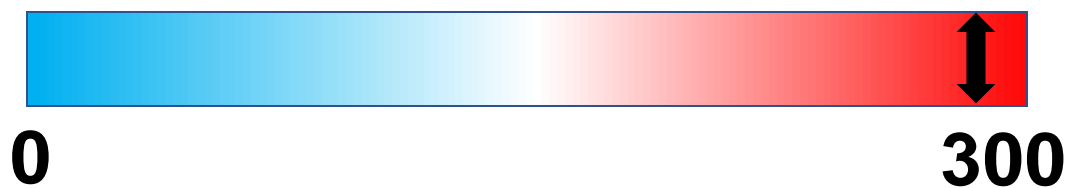
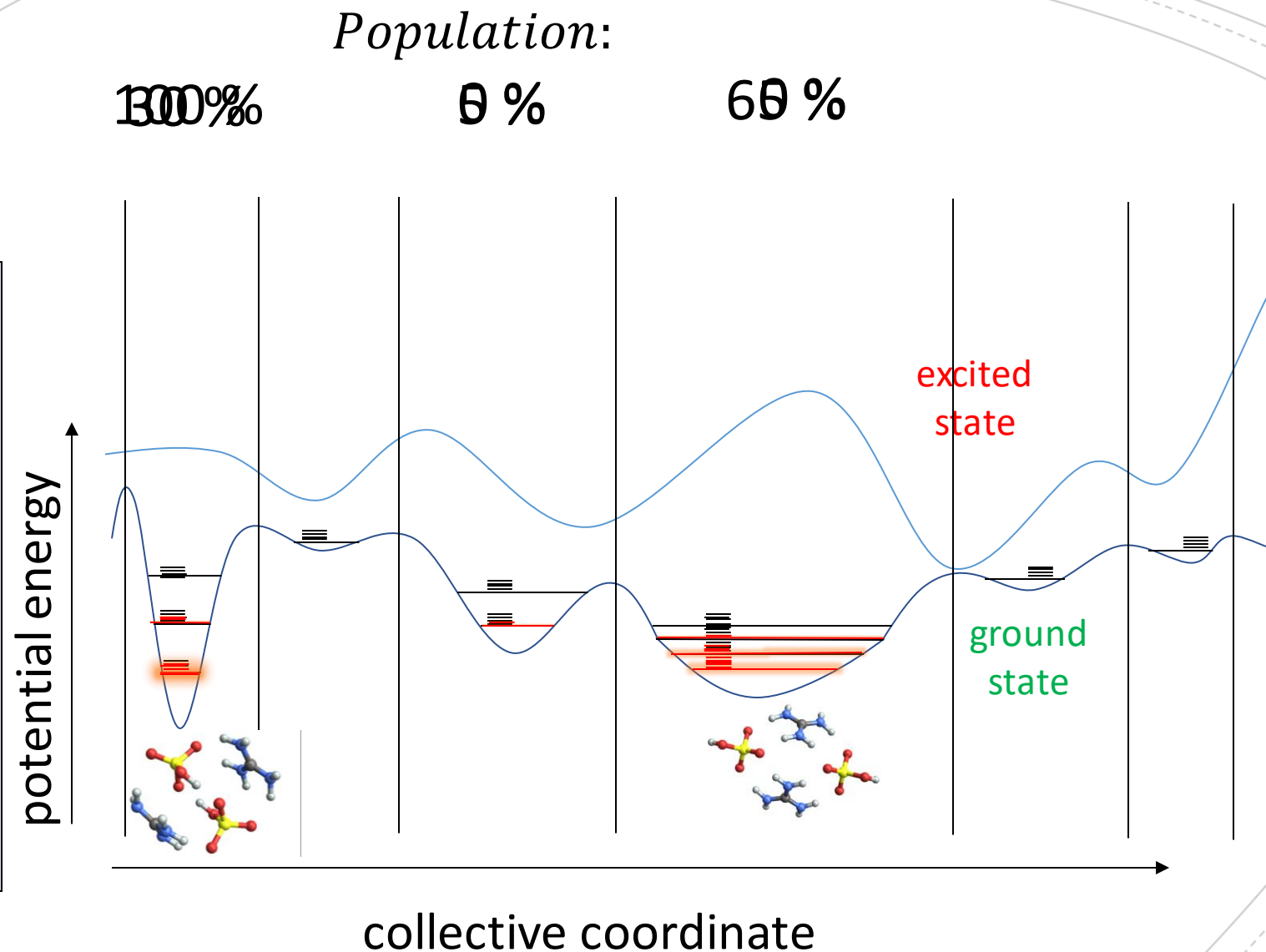


Configurational sampling

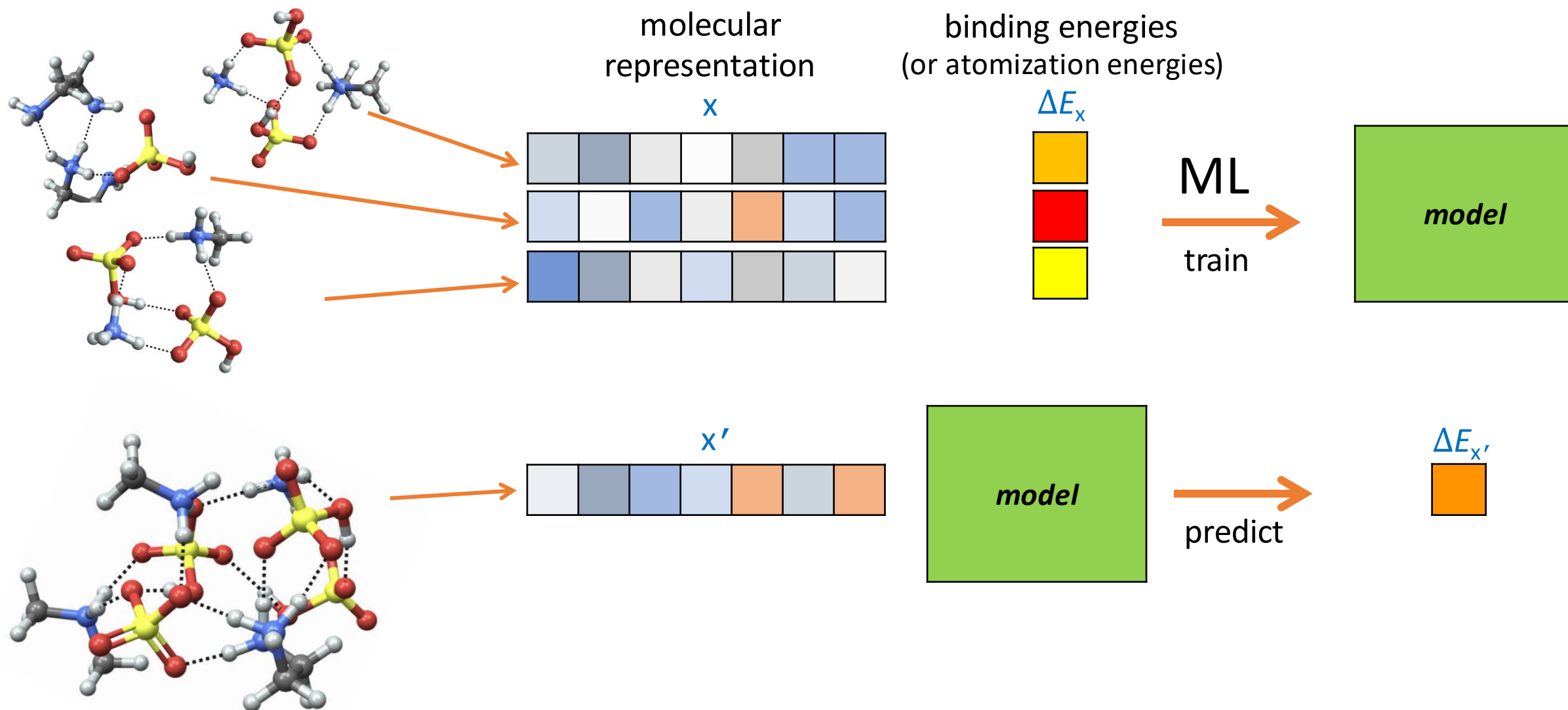
[Kubečka *et al.*, JPC, 2018]

Computational Tools for Molecular Clusters

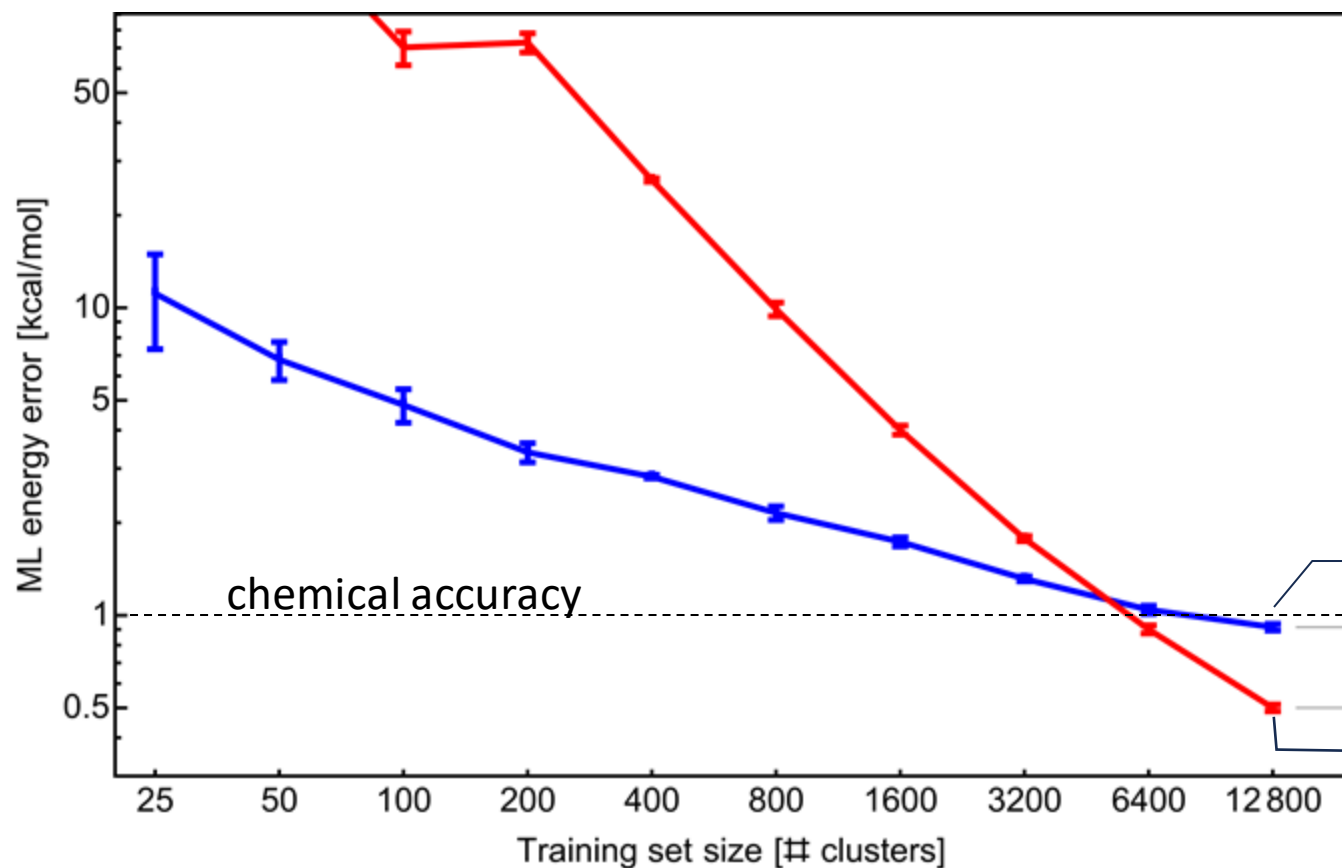
[Kubečka *et al.*, ACS Ω , 2023]



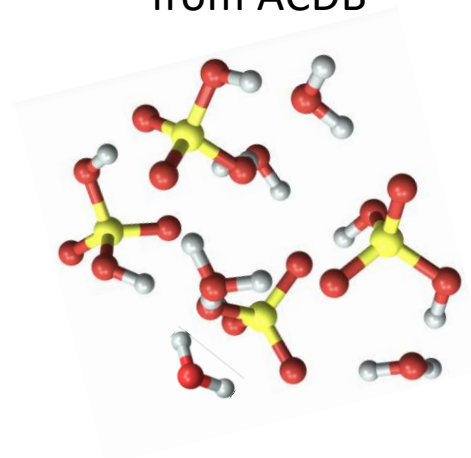
Machine Learning (e.g. GPR, KRR, or NN)



Accuracy of ML methods



Tested on 17k data
of sulfuric acid and water
from ACDB



7 days / 4 CPUs

KRR: QML/FCHL19

SchNetPack/PaiNN

8 hours / 4 CPUs + 1 GPU

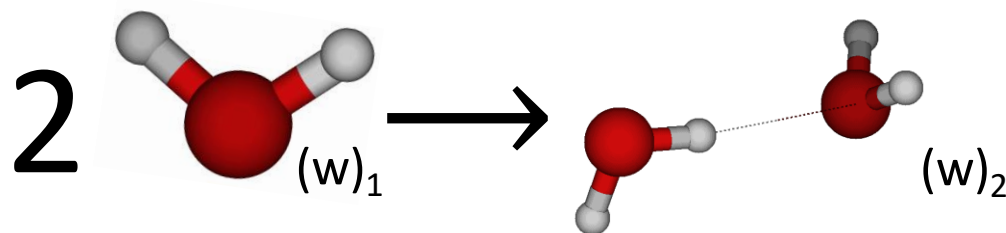
PREDICTION
4000 structures
just 3 min

Are the QC methods themselves accurate?!

DLPNO^{Normal}PNO-CCSD(T)/aug-cc-pVTZ// ω B97X-D/6-31++G(d,p)
+ vibrational anharmonicity scaling factor 0.996
+ low vibrational frequency treatment with cutoff of 100 cm⁻¹
+ monomer rotational symmetry correction
+ multi low-energy minima treatment

Reactional Free Energy

Water dimerization:

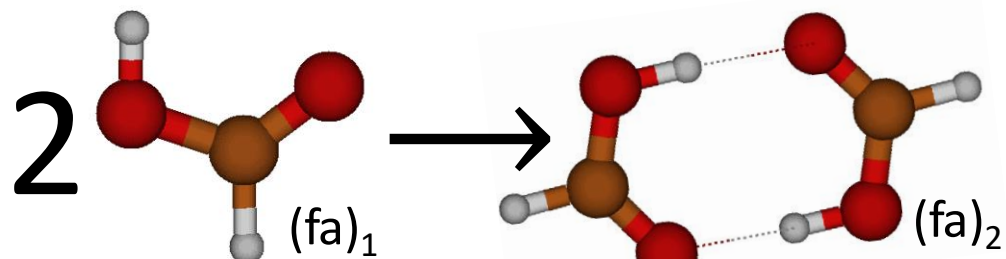


THEORY: **+3.19** kcal/mol

EXP: **+3.15** ± 0.12 kcal/mol

[Auwera et al., J.Chem.Phys. (2007)]

Formic acid dimerization:

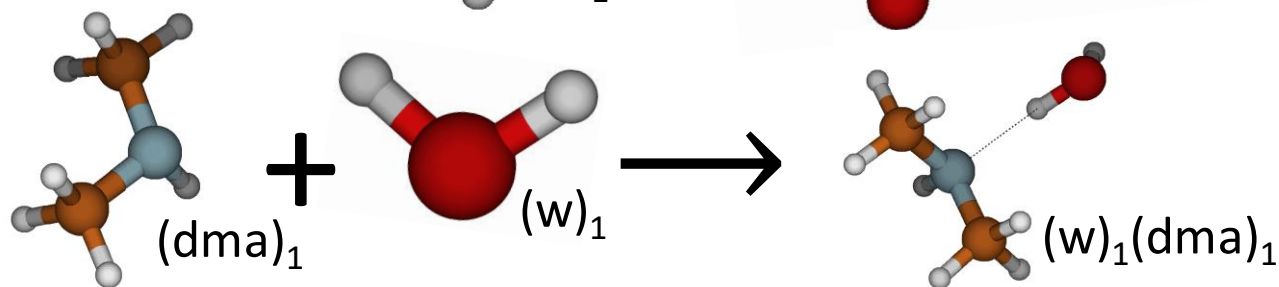


THEORY: **-3.33** kcal/mol

EXP: **-3.31** ± 0.04 kcal/mol

[Rocher-Casterline et al., J.Chem.Phys. (2011)]

Dimethylamine hydration:

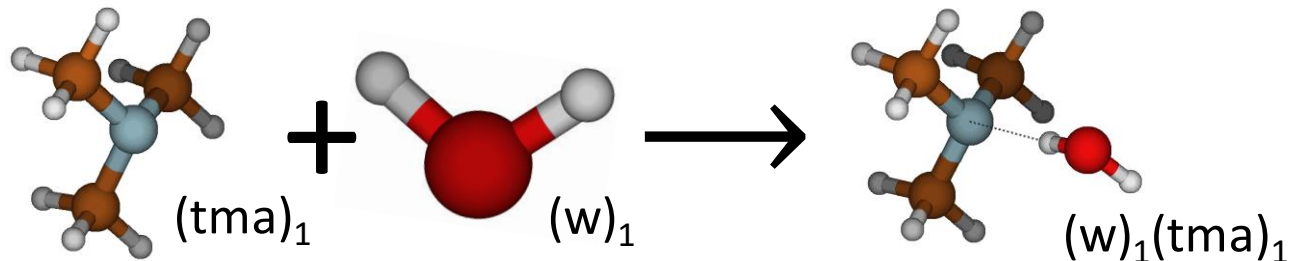


THEORY: **+1.91** kcal/mol

EXP: **+1.19** ± 0.04 kcal/mol

[Kjærsgaard et al., J.Phys.Chem.A (2020)]

Trimethylamine hydration:

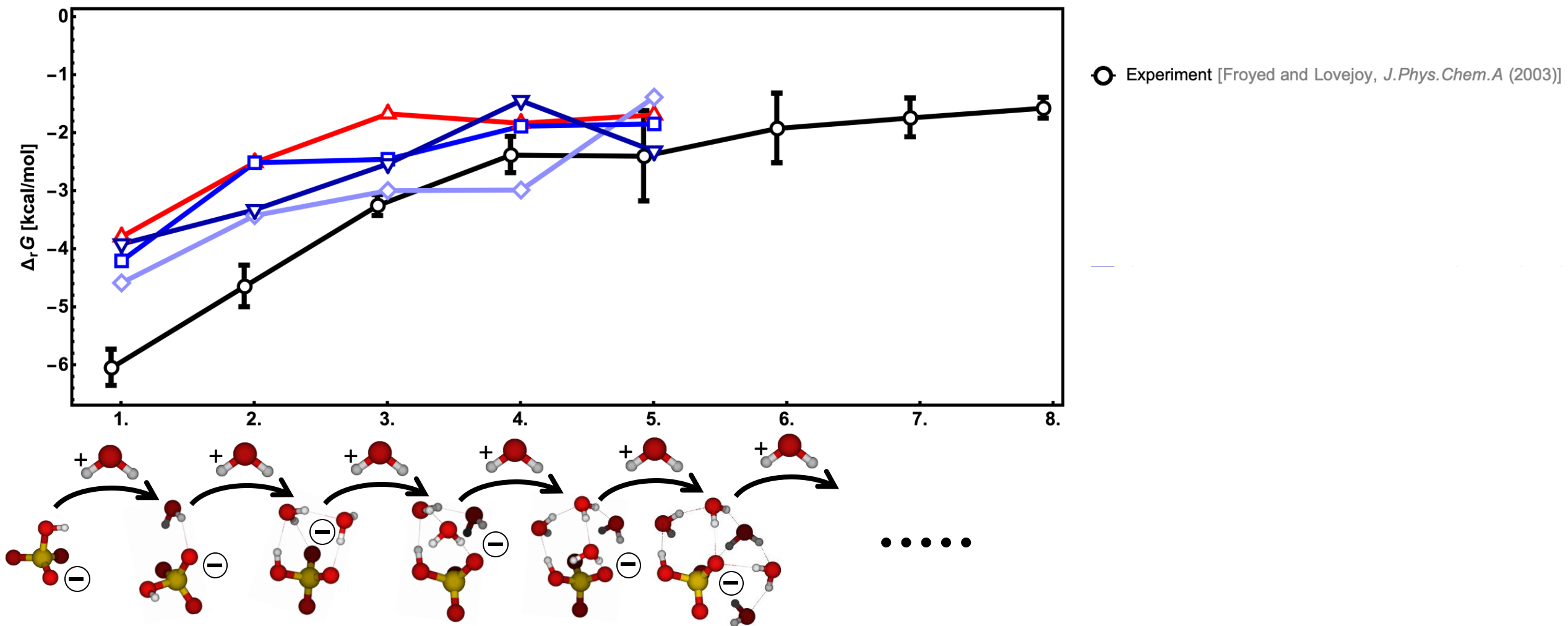


THEORY: **+1.30** kcal/mol

EXP: **+0.91** ± 0.04 kcal/mol

[Kjærsgaard et al., J.Phys.Chem.A (2020)]

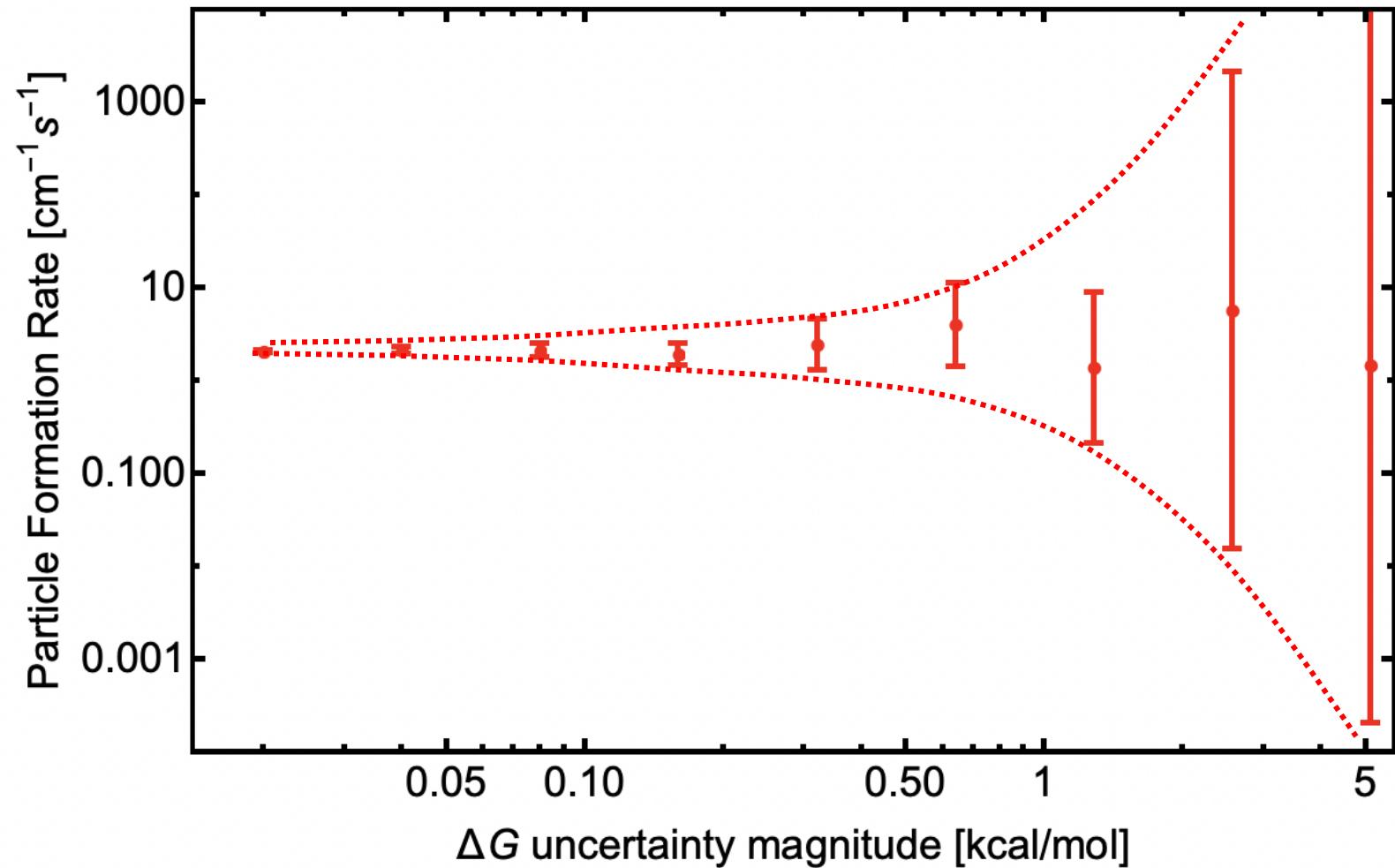
Bisulfate hydration



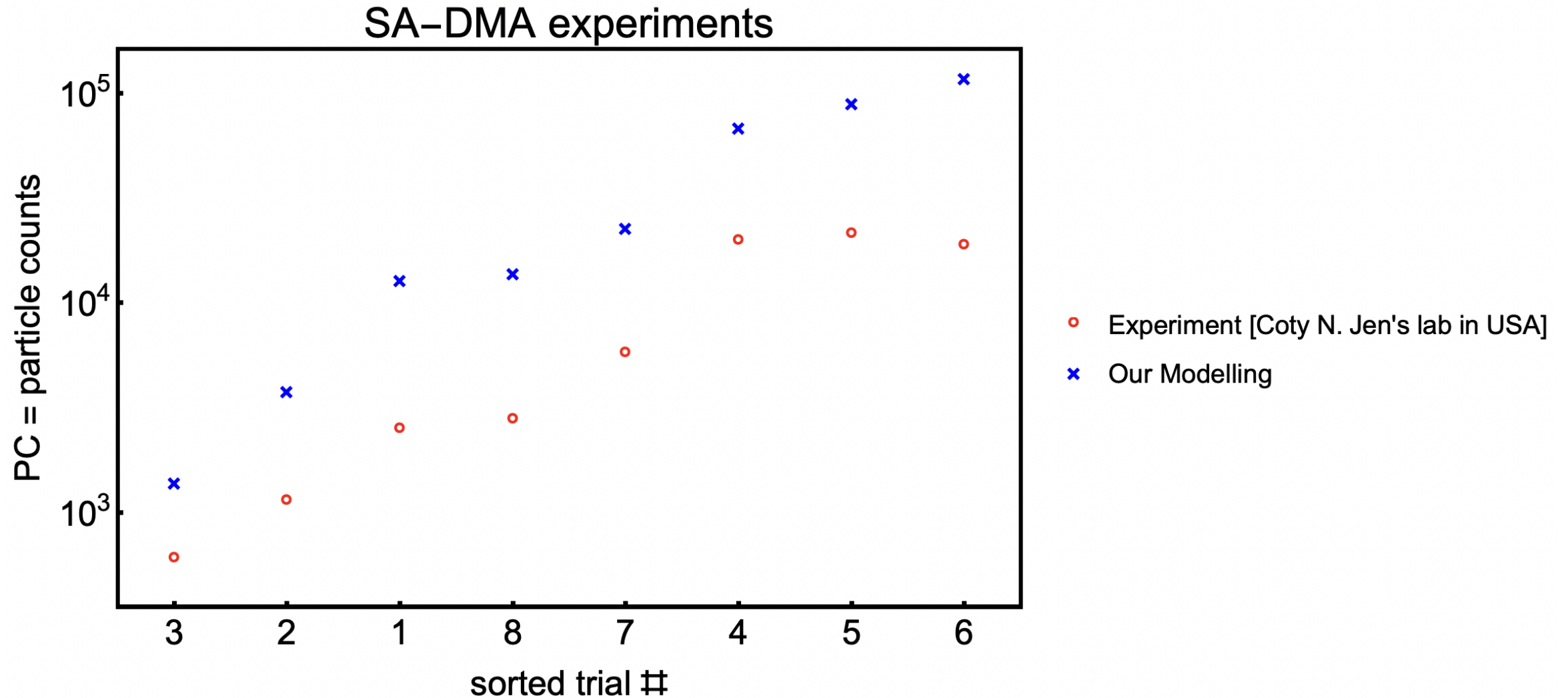
Potential errors in predictions

- ACDC simulation:
(ACDB data)

100 ppb NH_3
($\approx 10^{12} \text{ cm}^{-3} \text{ NH}_3$)
+
 $10^8 \text{ cm}^{-3} \text{ H}_2\text{SO}_4$
-----298.15 K-----
 $\sim 2 \text{ particles} \cdot \text{cm}^{-3} \cdot \text{s}^{-1}$

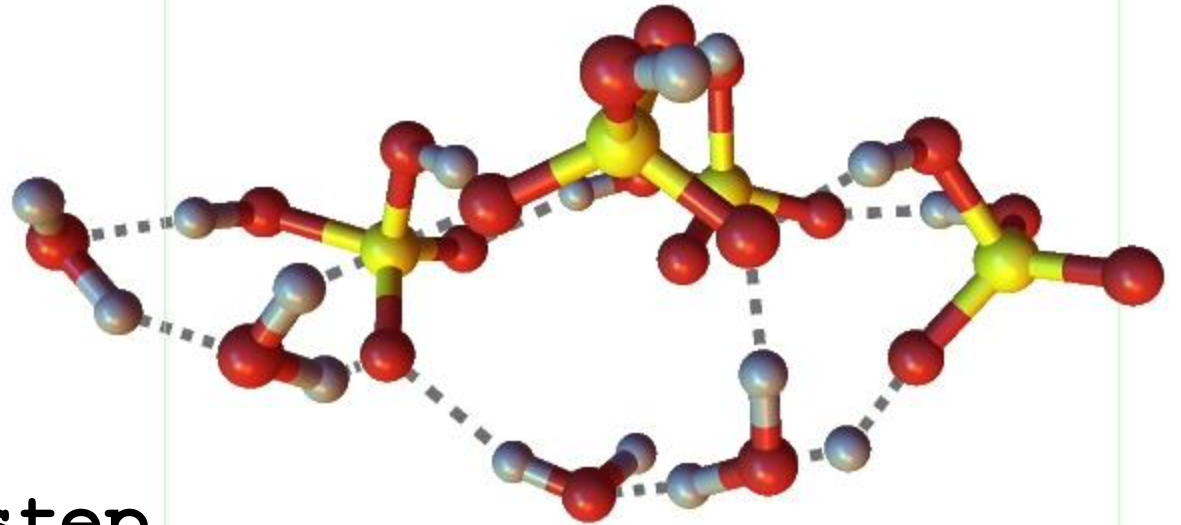


Systematic Error for Strongly Binding Clusters



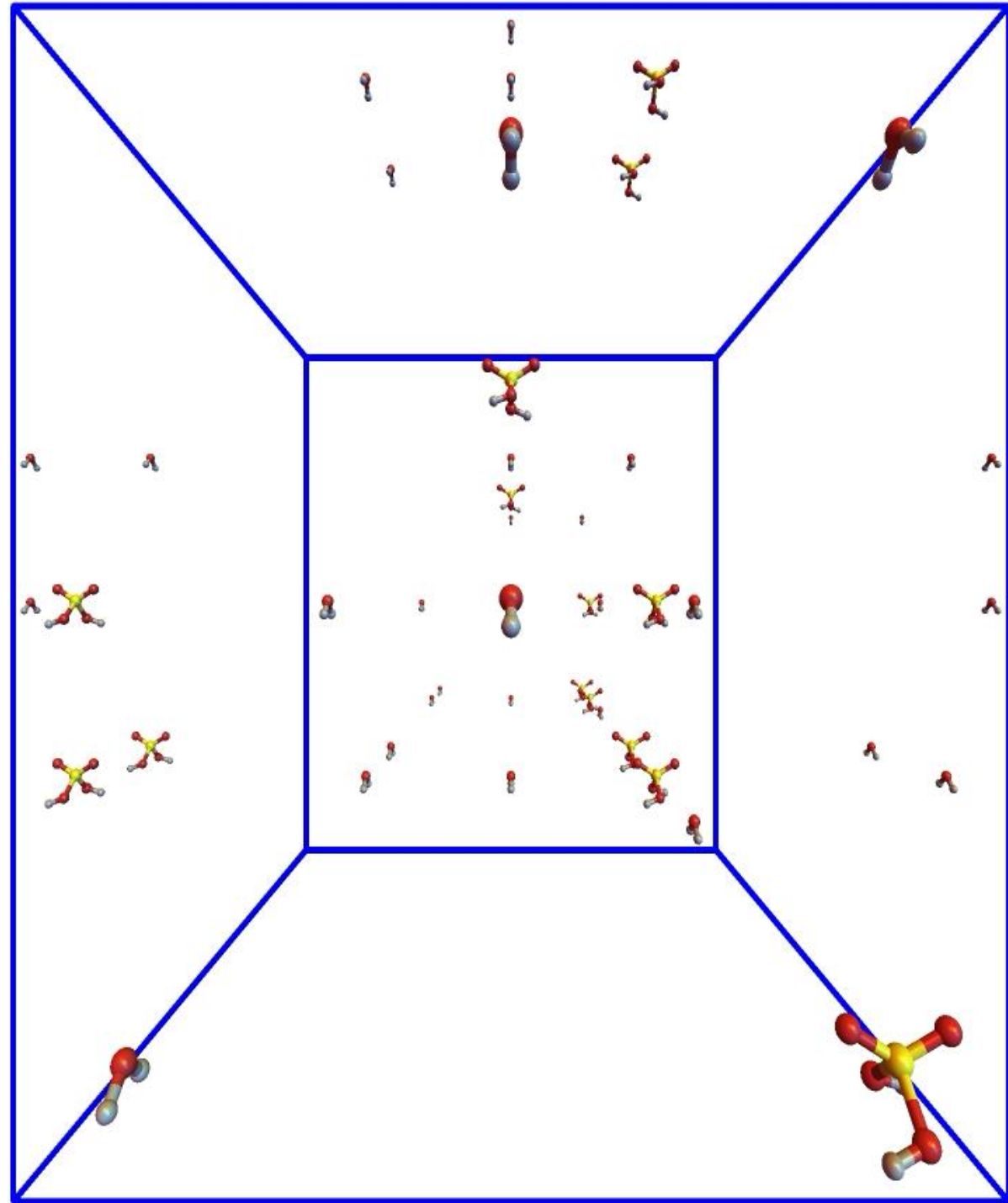
Simulating CLUSTER

- **Level = DFT via NN**
 ω B97X-D/6-31++G(d,p)
- **System = 4 × (H₂SO₄)**
4 × (H₂O)
- **MD = Langevin (300 K)**
1 fs integration step
0.1 ns simulation



Simulation BOX

- **Level = DFT via NN**
 ω B97X-D/6-31++G(d,p)
- **Box = $40 \times 40 \times 40 \text{ \AA}^3$**
- **System = $10 \times (\text{H}_2\text{SO}_4)$
 $10 \times (\text{H}_2\text{O})$**
- **MD = Langevin (300 K)**
1 fs integration step
1 ns simulation



Conclusion and Outlook

- Atmospheric NPF
- QC + Configurational Sampling of Molecular Clusters
- (Conformational) Entropy Needs To Be Addressed Now!

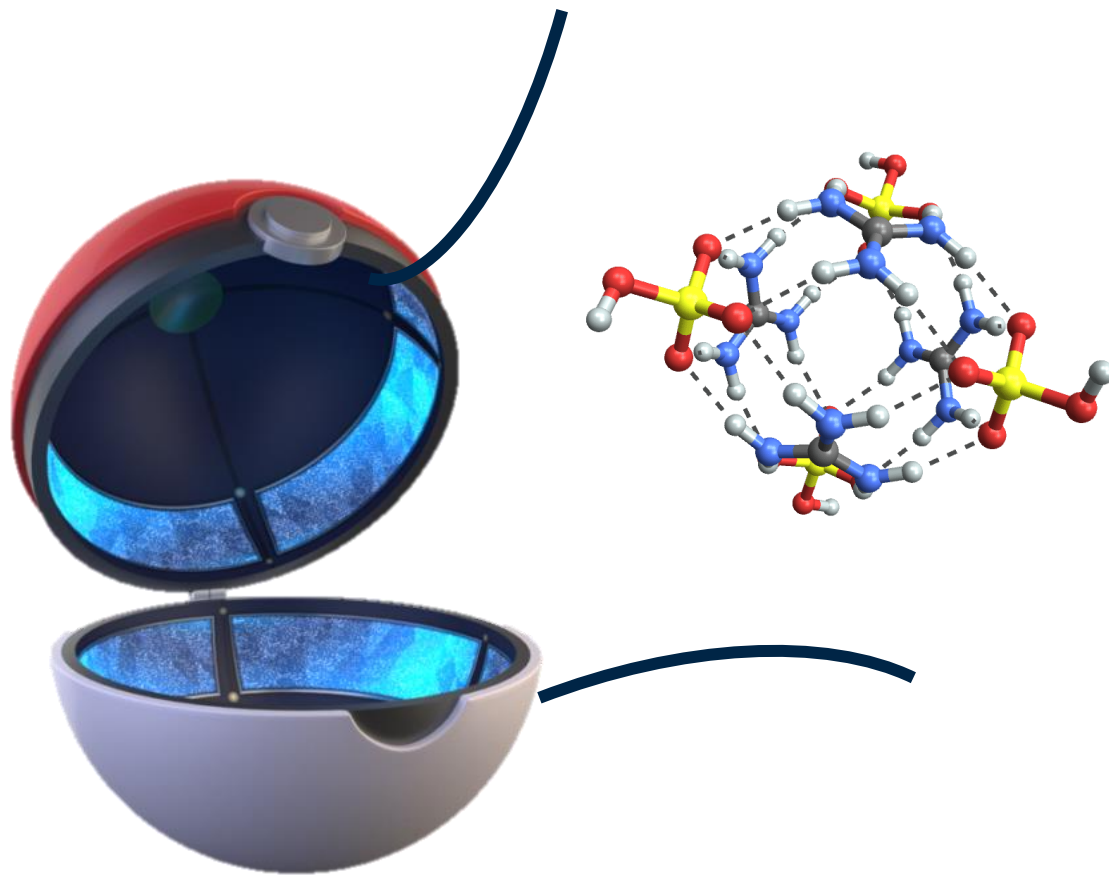
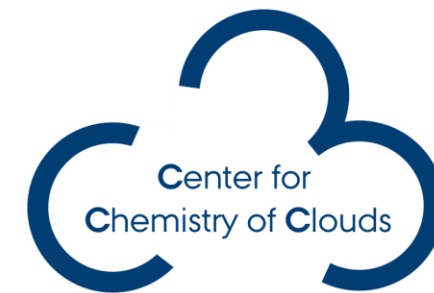
TAKE HOME MESSAGE:





Funded by the
European Union

Thank you



Gotta **CATCH**'em all!

CAtCh

Computational Atmospheric Chemistry



From left to right: Haide Wu, Daniel Ayoubi, Yosef Knattrup, Jakub Kubečka, Morten Engsvang, Jonas Elm, Andreas Buchgraitz Jensen, Jakob Lund, Astrid Nørskov Pedersen